153. A Search for New Trypanocides. Part IV.* Some Derivatives of 4-Amino-6-aminomethylquinaldine and 1: 2-Di-(4-aminoquinald-6-yl)ethylene.

By J. N. ASHLEY and M. DAVIS.

NN'-Di-(4-aminoquinald-6-ylmethyl)urea (Id), 1: 3-di-[(4-aminoquinald-6-ylmethyl)amino]propane (Ie), 2-amino-4: 6-di-[(4-aminoquinald-6-ylmethyl)amino]-1: 3: 5-triazine (If), 4-amino-6-[(4-aminoquinald-6-ylmethyl)-amino]quinaldine (Ig), 1: 2-di-(4-aminoquinald-6-yl)ethylene (Ih), 2: 3-di-(4-aminoquinald-6-yl)but-2-ene (Ii), and several of their derivatives and quaternary salts have been prepared as potential trypanocides.

In recent years, trypanocidal activity has been reported for many symmetrical quinoline derivatives of type (I) in which two 4-aminoquinaldine residues are joined at the 6-position by various chains or cyclic groups. Compounds most extensively examined include $\alpha\omega$ -di-(4-aminoquinald-6-ylamino)alkanes (Ia), $\alpha\omega$ -di(4-aminoquinald-6-yloxy)alkanes (Ib),



and the diamides (Ic) of 4:6-diaminoquinaldine with various dicarboxylic acids.¹ Many derivatives of 4-aminoquinaldine joined in the 6-position to heterocyclic compounds such as 2:4:6-triamino-1:3:5-triazine or 2:4-diamino-6-methylpyrimidine have exhibited high trypanocidal activity,² and symmetrical azo-compounds, ureas, thioureas, and

• Part III, J., 1956, 3739.

¹ Jensch, Z. angew. Chem., 1937, **50**, 891; Pratt and Archer, J. Amer. Chem. Soc., 1948, **70**, 4065; Goble, J. Pharmacol., 1950, **98**, 49.

² Barrett, Curd, and Hepworth, J., 1953, 50.

guanidines derived from 4:6-diaminoquinoline, and linked through the 6-position, have been studied by Keneford et al.³

We have sought to extend this work and have prepared a number of compounds containing two 4-aminoquinaldine nuclei connected in the 6-position by various linkages. The 2-methyl group in the quinoline nucleus has been retained in all the compounds synthesised since it has been shown to be essential for high activity.⁴ In one series of compounds derived from 1 : 2-di-(4-aminoquinald-6-yl)ethylene (Ih) the effect of replacing the 4-aminogroup by other substituents has also been examined.

It was decided at first to synthesise several homologues of known trypanocides employing 4-amino-6-aminomethylquinaldine in place of 4:6-diaminoquinaldine. The required quinaldine was readily obtained by the Conrad-Limpach method from N-4-aminobenzylacetamide, which was prepared by catalytic reduction of the known N-4-nitrobenzylacetamide. The resulting 6-acetamidomethyl-4-hydroxyquinaldine (II; R = Ac, R' =OH) was methylated to give the 4-methoxy-compound (cf. Pratt and Archer 1) and this was fused with ammonium acetate yielding, after hydrolysis, 4-amino-6-aminomethylquinaldine (II; R = H, $R' = NH_{2}$). The diamino-derivative reacted normally with carbonyl chloride, trimethylene dibromide, and 2-amino-4:6-dichloro-1:3:5-triazine, forming respectively NN'-di-(4-aminoquinald-6-ylmethyl)urea (Id), 1:3-di[(4aminoquinald-6-ylmethyl)amino]propane (Ie), and 2-amino-4:6-di-[(4-aminoquinald-6-ylmethyl)amino]-1:3:5-triazine (If). With methyl dichloroacetate, the dichloroacetyl derivative (II; $R = CO \cdot CHCl_2$, $R' = NH_2$) was formed. In the presence of sodium carbonate, methyl iodide gave the bisquaternary compound, 4-amino-6-dimethylaminomethylquinaldine bismethiodide. The monoquaternary salt, 4-amino-6-aminomethylquinaldine methiodide was produced when the 6-acetyl derivative was treated with methyl iodide and then hydrolysed with hydriodic acid.

Another compound prepared (Ig) consisted of a 4-amino-6-methylquinaldine and a 4:6diaminoquinaldine linked together through the 6-position. It is thus related to both the known trypanocides of type (Ia) and the homologous compound (Ie) described above. Condensation of 4-nitrobenzyl chloride with 4:6-diaminoquinaldine gave 4-amino-6-4'nitrobenzylaminoquinaldine (III; R = H, $R' = NO_2$) which was acetylated and then reduced catalytically to 4-acetamido-6-(N-4-aminobenzy) acetamido) quinaldine (III; R = Ac, $R' = NH_{2}$). Application of the Conrad-Limpach reaction to the latter base afforded 4-acetamido-6-[N-(4-hydroxyquinald-6-ylmethyl)acetamido]quinaldine (IV; R = R' =Ac, R'' = OH). The corresponding 4-chloro-compound (IV; R = R' = Ac, R'' = Cl) showed unusual stability. Thus, the 4-chloro-group was unaffected during the hydrolysis of the acetyl groups by 2n-hydrochloric acid at 100°, whilst treatment with alcoholic ammonia at 120° resulted only in the loss of one acetyl group, giving probably (IV; R =H, R' = Ac, R'' = Cl). However, replacement of the chloro-group was readily effected by heating the compound with ammonia in phenol.⁵ Hydrolysis of the product then gave the dihydrochloride of the required 4-amino-6-[(4-aminoquinald-6-ylmethyl)amino]quinaldine (Ig).

In the aromatic diamidine series, high trypanocidal activity is retained when a central chain •O•[CH₂]_n•O• is replaced by the unsaturated linkage, •CH:CH•.⁶ We therefore investigated the corresponding quinaldine (Ih). To this end, 4:4'-diaminostilbene was submitted to the usual Conrad-Limpach condensation and the resulting 1 : 2-di-(4-hydroxyquinald-6-yl)ethylene (V; R = OH, R' = H) was methylated and then fused with ammonium acetate, yielding 1 : 2-di-(4-aminoquinald-6-yl)ethylene (Ih). In order to study the effect of substitution in the amino-groups, the 4:4'-dichloro-compound (V; R = Cl, R' = H) was condensed with a number of primary and secondary amines to form the

³ Keneford, Lourie, Morley, Simpson, Williamson, and Wright, J., 1952, 2595.

⁴ Jensch, Annalen, 1950, **588**, 73.
⁵ Backenberg and Marais, J., 1942, 381; Albert, Brown, and Duewell, J., 1948, 1284.
⁶ Ashley, Barber, Ewins, Newbery, and Self, J., 1942, 103.

corresponding secondary and tertiary bases (see Table), most of which were converted into quaternary salts.

It has also been demonstrated in the diamidine series that replacement of the central CH:CH linkage by CMe:CMe causes a marked increase in activity against *T. congolense.*⁷ The related quinaldine (Ii) was therefore similarly prepared from acetoacetic ester and 2:3-di-(p-aminophenyl)but-2-ene,⁸ the final amination being effected in this case by ammonia-phenol treatment of the chloro-compound (V; R = Cl, R' = Me).

		1:2-D	<i>i-</i> (4-R	l-quinal	d-6-yl)	ethylen	es.				
No.	R Derivative		Conditions			Yield (%)		М. р.		Form	
1	NHMe	HMe Base ^a		150°; 16 hr.				340° *		Rect. plates	
2		2HCl									-
3		2Me₂SO₄		3 hr.		74				-	-
4	NHEt	Base •	150-	—160°;	20 hr.	79		320—3	25 *	Prisms	
5		2HCl		<u> </u>							-
6	NTE+	$2Me_2SO_4$ *	9 hr. 1959, 90 hr			52		>340		Blades	
1	NEt ₂ Base		Cu bronze added			14		102-104		Prisms	
8		2HCl ª						316-320 *		Prisms	
9		2Me ₂ SO ₄	2 hr.			64					
10	NH·[CH ₂] ₂ ·NEt ₂	NH·[CH ₂] ₂ ·NEt ₂ Base •		160°; 20 hr.			5	224-2	26	_	-
			Cu bronze added								
11		4HCl									
12	Dimenidime	$4Me_2SU_4$	3 hr.			78					
13	Piperidino	Pase -	185°; 7 nr.			80		242		Prisms	
15	_	2Me.SO.	3 hr			78	150-152		59	Plates	-
16	Piperazino	Base d	$185^{\circ}: 20 \text{ hr}$			71	So	Softens $> 240^{\circ}$		- A AALOG	-
17		4HCl							_	-	
			Found (%)			Required (%)					
			F	ound (%	o)			Re	quired	(%)	
No.	Formula	c	F H	ound (%	5) Cl	H ₂ O	c_	Re H	quired N	(%) Cl	H ₂ O
No. 1	Formula C24H24N4,H2O	C 74·5	F H 7·2	ound (% N 14·1	CI	H ₂ O	С 74·6	Re H 6·8	quired N 14·5	(%) Cl	H₂O
No. 1 2	Formula C ₂₄ H ₂₄ N ₄ ,H ₂ O C ₂₄ H ₂₄ N ₄ ,2HCl,H ₂	C 74·5 0 —	F H 7·2	ound (%) N 14·1 12·25	Cl 15.85	H ₂ O	C 74·6	H 6·8	quired N 14.5 12.2	(%) Cl 15·45	H₂O
No. 1 2 3	Formula C24H24N4,H2O C24H24N4,2HCl,H2 C28H34O4N4,2HCl,H2 C28H34O4N4S2	C 74·5 	F H 7·2 —	ound (% N 14·1 12·25 8·9	Cl 15-85	H ₂ O	C 74·6	H 6·8 —	quired N 14.5 12.2 9.0	(%) Cl 15·45	H₂O
No. 1 2 3 4	Formula C ₂₄ H ₂₄ N ₄ ,H ₂ O C ₂₄ H ₂₄ N ₄ ,2HCl,H ₂ C ₂₈ H ₃₆ O ₈ N ₄ S ₂ C ₂₆ H ₂₈ N ₄ O ₅ DH _{2O}	C 74·5 — 76·7	F H 7·2 	ound (% N 14·1 12·25 8·9 13·8	Cl 15-85	H ₂ O — — — —	C 74·6 77·0	Re H 6·8 — 7·2	quired N 14.5 12.2 9.0 13.8	(%) Cl 15·45	H ₂ O — — 2·2
No. 1 2 3 4 5	Formula C24H24N4,H2O C24H24N4,2HCl,H2 C28H36O4N4S2 C26H28N4,0.5H2O C26H28N4,0.5H2O C26H28N4,0.2HCl,H2	C 74.5 	F H 7·2 — 7·2 —	ound (% N 14·1 12·25 8·9 13·8 11·6	Cl 	H ₂ O — — 1·7	C 74·6 77·0 	H 6.8 — 7.2 —	quired N 14.5 12.2 9.0 13.8 11.5	(%) Cl 15·45 14·6	H ₂ O
No. 1 2 3 4 5 6 7	Formula C ₂₄ H ₂₄ N ₄ ,H ₂ O C ₂₄ H ₂₄ N ₄ ,2HCl,H ₂ C ₂₈ H ₃₆ O ₈ N ₄ S ₂ C ₂₆ H ₂₈ N ₄ ,0·5H ₂ O C ₂₆ H ₂₈ N ₄ ,2HCl,H ₂ C ₃₀ H ₄₀ O ₈ N ₄ S ₂	C 74·5 	F H 7·2 7·2 	ound (% N 14·1 12·25 8·9 13·8 11·6 8·8 12·2	Cl 15.85 14.6 	H ₂ O — — — — — — —	C 74·6 77·0 79.4	H 6.8 	quired N 14.5 12.2 9.0 13.8 11.5 8.6 12.4	(%) Cl 15·45 14·6 	H ₂ O — — 2·2 —
No. 1 2 3 4 5 6 7 8	Formula C ₂₄ H ₂₄ N ₄ , H ₂ O C ₂₄ H ₂₄ N ₄ , 2HCl, H ₂ C ₂₈ H ₃₆ O ₈ N ₄ S ₂ C ₂₆ H ₂₈ N ₄ , 2HCl, H ₁ C ₃₀ H ₄₀ O ₈ N ₄ S ₂ C ₃₀ H ₃₆ N ₄ C ₃₀ H ₃₆ N ₄ C ₃₀ H ₃₆ N ₄	C 74·5 	F H 7·2 7·2 8·0	ound (% N 14·1 12·25 8·9 13·8 11·6 8·8 12·2 10·9	Cl 15.85 14.6 13.35	H ₂ O	C 74·6 77·0 79·4	H 6·8 	quired N 14·5 12·2 9·0 13·8 11·5 8·6 12·4 10·7	(%) Cl 15·45 14·6 13·5	H ₂ O 2·2
No. 1 2 3 4 5 6 7 8 9	Formula C24H24N4,H2O C24H24N4,2HCl,H2 C28H36O8N4S2 C26H28N4,0-5H2O C26H28N4,0-2HCl,H2 C30H40O8N4S2 C30H36N4 C30H36N4,2HCl C3H36N4,2HCl C30H36N4,2HCl	C 74.5 ₀O — ₂O — ₃O — 79.4 —	F H 7·2 7·2 8·0	ound (% N 14·1 12·25 8·9 13·8 11·6 8·8 12·2 10·9 8·2	Cl 15·85 14·6 13·35	H ₂ O — — — — — — — — — — —	C 74·6 77·0 79·4 	Re H 6·8 	quired N 14·5 12·2 9·0 13·8 11·5 8·6 12·4 10·7 8·0	(%) <u>Cl</u> <u>15·45</u> <u></u> <u>14·6</u> <u></u> <u>13·5</u>	H ₂ O — 2·2 —
No. 1 2 3 4 5 6 7 8 9 10	Formula C24H24N4,H2O C24H24N4,2HCl,H2 C28H3603N3S2 C26H28N4,0-5H2O C26H28N4,0-5H2O C26H28N4,0-8N4S2 C30H36N4 C30H36N4,2HCl C30H36N4,2HCl C34H46O8N4S2 C34H46O8N4S2 C34H46N6	€ 74.•5 76.•7 20 — 79.•4 — 76.•1	F H 7·2 7·2 8·0 8·5	ound (% N 14·1 12·25 8·9 13·8 11·6 8·8 12·2 10·9 8·2 15·7	Cl 15.85 14.6 13.35 	H ₂ O	C 74·6 77·0 79·4 75·9	Re H 6·8 7·2 8·0 8·5	quired N 14.5 12.2 9.0 13.8 11.5 8.6 12.4 10.7 8.0 15.6	(%) <u>Cl</u> <u>15·45</u> <u>-</u> <u>14·6</u> <u>-</u> <u>13·5</u> <u>-</u>	H ₂ O
No. 1 2 3 4 5 6 7 8 9 10 11	Formula C ₂₄ H ₂₄ N ₄ , H ₂ O C ₂₄ H ₂₄ N ₄ , 2HCl, H ₂ C ₂₈ H ₃₆ O ₈ N ₄ S ₂ C ₂₆ H ₁₈ N ₄ , 0.5Hc ₂ O C ₂₆ H ₁₈ N ₄ , 0.5Hc ₂ O C ₃₀ H ₃₆ N ₄ S ₂ C ₃₀ H ₃₆ N ₄ C ₃₀ H ₃₆ N ₄ , 2HCl C ₃₄ H ₄₆ N ₆ , 2HCl	C 74·5 76·7 20 — 79·4 — 76·1	F H 7·2 7·2 8·0 8·5 8·5	ound (% N 14·1 12·25 8·9 13·8 11·6 8·8 12·2 10·9 8·2 15·7 12·4	Cl 	H ₂ O — — — — — — — — — — — — — — — — —	C 74·6 77·0 79·4 75·9 	Re H 6·8 	quired N 14·5 12·2 9·0 13·8 11·5 8·6 12·4 10·7 8·0 15·6 12·3	(%) <u>C1</u> <u>15·45</u> <u>115·45</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u> <u>111</u>	H ₂ O
No. 1 2 3 4 5 6 7 8 9 10 11 12	Formula C24H24N4,H2O C24H24N4,2HCl,H2 C28H28O3N4S2 C28H28N4,0-5H2O C28H28N4,2HCl,H2 C30H38N4,2HCl,H2 C30H38N4,2HCl C30H38N4,2HCl C34H48O6N4S2 C34H46N6 C34H46N6,4HCl C44H70O18N6S4	C 74·5 	F H 7·2 	ound (% N 14·1 12·25 8·9 13·8 11·6 8·8 12·2 10·9 8·2 15·7 12·4 8·3	Cl 15·85 14·6 13·35 20·75	H ₂ O — — — — — — — — — — — — — — — — — — —	C 74·6 77·0 79·4 75·9 	Re H 6·8 	quired N 14.5 12.2 9.0 13.8 11.5 8.6 12.4 10.7 8.6 12.3 8.1	(%) <u>Cl</u> <u></u>	H ₂ O
No. 1 2 3 4 5 6 7 8 9 10 11 12 13	Formula C24H24N4,H2O C24H24N4,2HCl,H2 C28H36O3N4S2 C26H28N4,0-5H2O C26H28N4,2HCl,H2 C30H38N4,2HCl,H2 C30H38N4,2HCl C30H38N4,2HCl C34H48O6N4S2 C34H46N6 C34H46N6 C34H46N6 C34H46N6 C34H46N6 C34H46N6 C34H46N6 C34H48N6 C34H4	C 74·5 76·7 20 — 79·4 — 76·1 — 77·95	F H 7·2 	ound (% N 14·1 12·25 8·9 13·6 11·6 8·8 12·2 10·9 8·2 15·7 12·4 8·3 11·3	Cl 15·85 14·6 13·35 20·75 	H ₂ O 	C 74·6 — 77·0 — 79·4 — 75·9 — 78·0	Re H 6·8 	quired N 14.5 12.2 9.0 13.8 11.5 8.6 12.4 10.7 8.0 15.6 12.3 8.1 11.4	(%) Cl 	H ₂ O — 2·2 — — — — — — — — — — — — — — — — —
No. 1 2 3 4 5 6 7 8 9 10 11 12 13 14	Formula C24H24N4,H2O C24H24N4,2HCl,H2 C28H36O3N4S2 C26H38O3N4S2 C26H38N4,0-5H2O C26H38N4,2HCl,H2 C30H38N4,2HCl C30H38N4,2HCl C30H38N4,2HCl C34H46N6,4HCl C4H70018N6S4 C32H46N4,H2O C32H38N4,H2O C32H38N4,H2O C32H38N4,H2O C32H38N4,H2O C32H38N4,H2O	$\begin{array}{c} & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ &$	F H 7·2 	ound (% N 14·1 12·25 8·9 13·8 11·6 8·8 12·2 10·9 8·2 15·7 12·4 8·3 11·3 9·6	Cl 	H ₂ O 	C 74·6 — 77·0 — 79·4 — 75·9 — 78·0 —	Re H 6·8 7·2 8·0 8·5 7·8	quired N 14.5 12.2 9.0 13.8 8.6 12.4 10.7 8.0 15.6 12.3 8.1 11.4 9.9	(%) Cl 	H ₂ O
No. 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16	Formula $C_{24}H_{24}N_4, H_2O$ $C_{24}H_{24}N_4, 2HCl, H_2$ $C_{28}H_{36}O_8N_8S_2$ $C_{26}H_{28}A_4, 0.5H_2O$ $C_{26}H_{28}N_4, 2HCl, H_1$ $C_{30}H_{30}N_4$ $C_{30}H_{36}N_4, 2HCl$ $C_{34}H_{46}O_8N_4S_2$ $C_{34}H_{46}O_8A_5S_2$ $C_{34}H_{46}N_6$ $C_{34}H_{46}O_8A_4HCl$ $C_{34}H_{46}O_8A_4HCl$ $C_{32}H_{36}N_4, 2HCl, H_2$ $C_{32}H_{36}N_4, 2HCl, H_2$	$\begin{array}{c} & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\$	F H 7·2 	ound (% N 14·1 12·25 8·9 13·8 11·6 8·8 12·2 10·9 8·2 15·7 12·4 8·3 11·3 9·6 7·8 16·9	Cl 	H ₂ O 	C 74·6 77·0 79·4 75·9 78·0 72.6	Re H 6·8 	quired N 14.5 12.2 9.0 13.8 11.5 8.6 12.4 10.7 8.0 15.6 12.6 12.6 12.6 12.4 10.7 8.0 15.6 12.3 8.1 11.4 9.9 7.7	(%) <u>Cl</u> <u></u> 15·45 <u></u> 14·6 <u></u> 13·5 <u></u> 20·8 <u></u> 12·5 <u></u>	H ₂ O
No. 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 17	Formula C24H24N4,H2O C24H24N4,2HCl,H2 C28H36O8N4S2 C26H28N4,0-5H2O C26H28N4,2HCl,H2 C30H36N4 C30H36N4,2HCl,H2 C30H36N4 C30H36N4,2HCl C34H46O8N4S2 C34H46N6 C34H46N64HCl C44H4004N4S2 C32H36N4,2HCl,H2 C32H46N6 C32H46N6 C32H36N4,2HCl,H2 C32H36N4,2HCl,H2 C32H36N4,2HCl,H2 C32H46N6 C32H36N4,2HCl,H2 C32H36N4,2HCl,H2 C32H36N4,2HCl,H2 C32H46N6 C32H46N6 C32H46N6 C32H46N6 C32H36N4,2HCl,H2 C32H36N4,2HCl,H2 C32H36N4,2HCl,H2 C32H46N6 C32H46N6 C32H46N6 C32H46N6 C32H46N6 C32H36N4,2HCl,H2 C32H46N6 C32H36N4,2HCl,H2 C32H46N6 C32H36N4,2HCl,H2 C32H36N4,2HCl,H2 C32H46N6 C32H36N4,2HCl,H2 C32H46N6 C32H36N4,2HCl,H2 C32H36N4,2HCl,H2 C32H46N6 C32H36N4,2HCl,H2 C32H46N6 C32H46N6 C32H36N4,2HCl,H2 C32H36N6 C32H36	$\begin{array}{c} & & & \\ & & & & \\ & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\$	F H 7·2 	ound (% N 14·1 12·25 8·9 13·8 11·6 8·8 12·2 10·9 8·2 15·7 12·4 8·3 11·3 9·6 7·8 16·2 12·4	Cl 	H ₂ O 	C 74·6 77·0 79·4 75·9 78·0 72·6	Re H 6·8 	quired N 14.5 12.2 9.0 13.8 8.6 12.4 10.7 8.0 15.6 12.3 8.1 11.4 9.9 7.7 16.9 9.1 2.3	(%) Cl 	H ₂ O

^a Recryst. from EtOH. ^b Recryst. from light petroleum (b. p. 40–60°). ^c Recryst. from $C_{6}H_{6}$. ^d Recryst. from CHCl₂. ^e Recryst. from dimethylformamide. ^{*} With decomp.

These compounds were examined in our Biological Research Laboratories, and it was found that, although some of the compounds effected cures in mice infected with T. rhodesiense and one or two were active against T. congolense, none was as active as any of the usually employed trypanocidal agents.

EXPERIMENTAL

N-4-Nitrobenzylacetamide (cf. Amsel and Hofman⁹) — N-Benzylacetamide (190 g.) was slowly added to fuming nitric acid (d 1.5; 475 ml.) which was stirred and cooled in a freezing

⁷ Fulton and Yorke, Ann. Trop. Med. Parasitol., 1942, **36**, 131; 1943, **37**, 80, 152; Wien, Brit. J. Pharmacol., 1946, **1**, 65.

⁸ Allen and Corwin, J. Amer. Chem. Soc., 1950, 72, 114.

⁹ Amsel and Hofmann, Ber., 1886, **19**, 1284.

mixture, so that the temperature did not exceed 35° . The solution was kept for 10 min., ice (950 g.) was added, and concentrated aqueous ammonia was introduced slowly, the temperature being kept at $<25^{\circ}$, until the mixture was alkaline to phenolphthalein. The precipitate was filtered off and recrystallised from water (charcoal), giving 128 g. (52%) of product, m. p. 130—131°.

N-4-Aminobenzylacetamide.—N-4-Nitrobenzylacetamide (41 g.) in methanol containing platinum oxide (1.5 g.) was reduced catalytically. The filtered solution was evaporated, and the residue was triturated with ether-light petroleum (b. p. 40—60°), to give N-4-aminobenzylacetamide (34.1 g., 98%; m. p. 90—96°) which separated from benzene or from chloroform-ether in plates, m. p. 99—101° (Found : N, 17.1. $C_9H_{19}ON_9$ requires N, 17.2%).

Ethyl γ -p-Acetamidomethylanilinocrotonate.—N-4-Aminobenzylacetamide (56 g.) and ethyl acetoacetate (56 g.) in ethanol (280 ml.) containing a few drops of concentrated hydrochloric acid were refluxed on the steam-bath overnight and then set aside. The crotonate (19·1 g.) which separated crystallised from ethyl acetate and aqueous methanol in needles, m. p. 113° (Found : C, 65·35; H, 7·1; N, 10·3. C₁₅H₂₀O₃N₂ requires C, 65·25; H, 7·3; N, 10·15%). Evaporation of the original mother-liquors and repeated crystallisation of the residue from methanol and from ethyl acetate gave a further quantity of the crotonate (total yield 39·9 g.), m. p. >110°. It is important that the crotonate be reasonably pure before any attempt is made to cyclise it.

6-Acetamidomethyl-4-hydroxyquinaldine.—Ethyl p-acetamidomethylanilinocrotonate (46 g.) was added during several minutes to stirred boiling Dowtherm (300 ml.), and boiling was continued for a further 10 min. After being cooled, the mixture was filtered, and the product was washed with benzene and purified by trituration with a little warm alcohol, cooling, and dilution with ether (yield 68%); it was pure enough for methylation. The pure product crystallised from ethanol in plates, m. p. 277° (Found : N, 12·15. C₁₃H₁₄O₂N₃ requires N, 12·2%).

6-Acetamidomethyl-4-methoxyquinaldine.—Finely powdered 6-acetamidomethyl-4-hydroxyquinaldine (17 g.) and methyl sulphate (10·2 g.) were stirred under reflux in boiling dry toluene (100 ml.) for 3 hr. After being cooled, the toluene was decanted and the residue washed with ether, dissolved in hot water (100 ml.), and basified with 2N-sodium hydroxide. The 6-acetamidomethyl-4-methoxyquinaldine was filtered off and recrystallised from boiling water (charcoal) in needles (11·85 g., 66%), m. p. 170—172° (Found : N, 11·3. $C_{14}H_{16}O_2N_2$ requires N, 11·5%).

6-Acetamidomethyl-4-aminoquinaldine.—The 4-methoxy-compound (11.85 g.) and ammonium acetate (60 g.) were heated at 140° for 3 hr. The clear solution was diluted with a little water and basified with concentrated sodium hydroxide solution, and the precipitate was filtered off and washed with water. Recrystallisation from water (charcoal) gave 6-acetamidomethyl-4-aminoquinaldine (9.1 g., 82%) in plates, m. p. 239—240° (Found : C, 67.9; H, 6.6; N, 17.9. C₁₃H₁₅ON₃ requires C, 68.0; H, 6.6; N, 18.3%). The methiodide, prepared by using methyl iodide in boiling methanol, crystallised from methanol in prisms, m. p. 264—266° (decomp.) (Found : N, 11.3; I, 33.8. C₁₄H₁₈ON₃I requires N, 11.3; I, 34.2%).

4-Amino-6-aminomethylquinaldine.—6-Acetamidomethyl-4-aminoquinaldine (9.1 g.) was heated with concentrated hydrochloric acid (70 ml.) in water (30 ml.) on the steam-bath overnight. The solution was cooled and the *dihydrochloride* (9.5 g., 92%) was filtered off and washed with alcohol; it crystallised from dilute hydrochloric acid-acetone in needles, m. p. >360° (Found : N, 15.7; Cl, 26.4; H₂O, 1.8. C₁₁H₁₃N₃,2HCl,0.25H₃O requires N, 15.9; Cl, 26.8; H₂O, 1.7%). The free base crystallised from chloroform in prisms, m. p. 215—216° (Found : C, 70.6; H, 6.8; N, 22.3. C₁₁H₁₃N₃ requires C, 70.6; H, 7.0; N, 22.45%). The 6-dichloro-acetyl derivative was prepared by boiling a solution of the base with methyl dichloroacetate in alcohol for 1 hr.; recrystallised from chloroform, it had m. p. 215—217° (Found : N, 14.3; Cl, 23.1. C₁₃H₁₃ON₃Cl₂ requires N, 14.1; Cl, 23.8%).

4-Amino-6-aminomethylquinaldine Methiodide Hydriodide.—6-Acetamidomethyl-4-aminoquinaldine methiodide (2 g.) was heated with 25% hydriodic acid (15 ml., freshly distilled) on the steam-bath for 20 hr., then cooled and filtered, and the residue washed with acetone. The methiodide hydriodide crystallised from methanol in narrow prisms, m. p. 279—281° (Found : N, 8.7; I, 53.4. $C_{12}H_{16}N_{3}I$,HI,H₂O requires N, 8.8; I, 53.4%).

Methylation of 4-Amino-6-aminomethylquinaldine.—The base $(1\cdot12 \text{ g.})$ was boiled with methyl iodide (30 ml.) and sodium carbonate $(0\cdot636 \text{ g.})$ in methanol (30 ml.) for 22 hr. The precipitate $(2\cdot55 \text{ g.})$ was filtered off and crystallised from methanol, yielding 4-amino-6-dimethylamino-methylquinaldine dimethiodide $(2\cdot2 \text{ g.})$, m. p. 276—278° (Found : C, 34.8; H, 4.5; N, 7.9; I,

50.8; N-Me, 11.2. $C_{18}H_{38}N_{3}I_{3}$, $H_{4}O$ requires C, 34.8; H, 4.9; N, 8.1; I, 49.1; N-Me, 11.6%). In absence of sodium carbonate, methylation was incomplete.

1: 3 - Di - [(4 - aminoquinald - 6 - ylmethyl)amino]propane. - 4 - Amino-6 - aminomethylquinaldine (1.55 g.) and trimethylene dibromide (0.84 g.) were heated in alcohol (20 ml.) in a sealed tube at 120° for 3 hr. The contents of the tube were evaporated and the residue treated with aqueous sodium hydroxide. The gummy base which separated slowly hardened. It was filtered off, washed with water, and dissolved in alcohol, the resulting solution (after clarification with charcoal) being acidified with dilute hydrochloric acid and then evaporated to dryness under reduced pressure. Trituration of the residue with warm alcohol gave the hygroscopic 1: <math>3 - di - [(4 - aminoquinald - 6 - ylmethyl)amino]propane tetrahydrochloride, decomp. > 265° (Found : N, 12·2; Cl, 20·4; H₂O, 18·1. C₂₅H₃₀N₆,4HCl,7H₃O requires N, 12·2; Cl, 20·7; H₄O, 18·4%). The amorphous free base had m. p. 90—110° (Found : C, 66·7; H, 7·5; N, 18·35. C₂₅H₃₀N₆,2H₃O requires C, 66·6; H, 7·6; N, 18·6%).

NN'-Di-(4-aminoquinald-6-ylmethyl)urea.—Carbonyl chloride was passed for 1.25 hr. through a solution of 4-amino-6-aminomethylquinaldine (1.5 g.) and sodium acetate (1.5 g.) in hot water (50 ml.). The clear solution was then cooled and basified with ammonia, and the *urea* filtered off and washed with water. Recrystallisation from a large volume of water gave needles (0.93 g.), decomp. >255° (Found: C, 55.9; H, 6.3; N, 17.5. $C_{23}H_{34}ON_{6.5}H_{3}O$ requires C, 56.3; H, 7.0; N, 17.1%). The dihydrochloride formed hygroscopic needles, m. p. >360°, from a large volume of alcohol (Found: N, 17.0; Cl, 14.9; H₂O, 3.7. $C_{23}H_{34}ON_{6.2}HCl,H_{3}O$ requires N, 17.1; Cl, 14.5; H₃O, 3.7%).

2-Amino-4: 6-di-[(4-aminoquinald-6-ylmethyl)amino]-1: 3: 5-triazine.—2-Amino-4: 6-dichloro-1: 3: 5-triazine (0.495 g.) was added to a solution of 4-amino-6-aminomethylquinaldine (1.12 g.) in nitrobenzene (10 ml.) at 160°. A precipitate commenced to separate almost immediately and, after being kept for 45 min. at 150—160°, the suspension was cooled, diluted with ether, and filtered, and the 2-amino-4: 6-di-[(4-aminoquinald-6-ylmethyl)amino]-1: 3: 5triazine dihydrochloride was recrystallised from alcohol, to yield 1.1 g., m. p. 270—280° (Found: N, 22.0; Cl, 11.1; H₂O, 14.9. C₂₅H₂₆N₁₀,2HCl,5H₂O requires N, 22.2; Cl, 11.3; H₂O, 14.3%).

4-Amino-6-4'-nitrobenzylaminoquinaldine.—4: 6-Diaminoquinaldine (20 g.) and 4-nitrobenzyl chloride (20 g.) in dry alcohol (200 ml.) were boiled under reflux for 6 hr. The solution was treated with water (400 ml.), 2N-hydrochloric acid (100 ml.), and charcoal, and then boiled. The insoluble residue left after filtration of the hot mixture was re-extracted with boiling water. Treatment of the combined extracts with 2N-sodium hydroxide and crystallisation of the precipitate from methanol gave 4-amino-6-4'-nitrobenzylaminoquinaldine (15.75 g.), orange rhombs, m. p. 199° (decomp.) (from chloroform) (Found : C, 62-0; H, 5.5; N, 17-2; H₂O, 7-0. $C_{17}H_{16}O_{2}N_{4}$, 1-25H₂O requires C, 61-7; H, 5.6; N, 16.9; H₂O, 6.9%). The hydrochloride crystallised from aqueous alcohol in yellow plates, m. p. 265—270° (decomp.) (Found : Cl, 10.6. $C_{17}H_{16}O_{2}N_{4}$, HCl requires Cl, 10.35%). The diacetyl derivative crystallised from chloroformbenzene in pale yellow prisms, m. p. 234—236° (Found : C, 65-1; H, 5-1. $C_{21}H_{20}O_{4}N_{4}$ requires C, 64-3; H, 5-15%).

4-Amino-6-4'-aminobenzylaminoquinaldine.—The above nitro-compound (2 g.) was reduced over platinum oxide in methanol at atmospheric temperature and pressure. Evaporation of the solution and crystallisation of the residue from chloroform gave the amine (1 g.) as colourless plates, m. p. 205—206° (Found : C, 73.2; H, 6.6. $C_{17}H_{18}N_4$ requires C, 73.4; H, 6.5%).

4-Acetamido-6-N-4'-aminobenzylacetamidoquinaldine.—4-Acetamido-6-N-4'-nitrobenzylacetamidoquinaldine (15.65 g.) was reduced over platinum oxide in methanol, at atmospheric temperature and pressure, and the residue obtained on evaporation of the solvent was repeatedly extracted with hot ethyl acetate. Concentration of the extracts gave 4-acetamido-6-N-4'-aminobenzylacetamidoquinaldine (10.1 g., m. p. >214°), colourless prisms, m. p. 221° (effervesces and loses solvent at 120—124°), from ethyl acetate [Found (after drying) : C, 70.0; H, 6.4; N, 15.45. $C_{g1}H_{2g}O_{g}N_{4}$ requires C, 69.7; H, 6.1; N, 15.5%].

Ethyl γ -p-[N-(4-Acetamidoquinald-6-yl)acetamidomethyl]anilinocrotonate.—The foregoing amino-compound (4.5 g.), ethyl acetoacetate (2 g.), and concentrated hydrochloric acid (1 drop), in alcohol (80 ml.), were heated under reflux for 18 hr. After distillation of the solvent, the residue was heated on the steam-bath for 1 hr., then extracted several times with boiling benzene (total 500 ml.), and the combined extracts were cooled and chromatographed over alumina. Elution with benzene gave the crotonate (3.15 g.; m. p. >135°), which formed colourless plates, m. p. 142—144°, from ethyl acetate-light petroleum (b. p. 40—60°) (Found : C, 68.3; H, 6.25; N, 11.9. $C_{27}H_{30}O_4N_4$ requires C, 68.3; H, 6.4; N, 11.8%). Further elution of the column with chloroform and acetone and purification of the original benzene-insoluble residue gave unchanged amino-compound (1.2 g.).

4-Acetamido-6-[N-(4-hydroxyquinald-6-ylmethyl)acetamido]quinaldine.—The above crotonate (7 g.) was slowly added with stirring to boiling Dowtherm (80 ml.). After 10 min. the solution was cooled, diluted with light petroleum (b. p. 40–60°), and filtered. The amorphous hydroxy-quinaldine could not be obtained crystalline. The *picrate* formed needles, m. p. 160–162°, from methanol (Found : N, 15·1. $C_{25}H_{24}O_3N_4, C_6H_3O_7N_3$ requires N, 14·9%).

4-Acetamido-6-[N-(4-chloroquinald-6-ylmethyl)acetamido]quinaldine.—The crude amorphous hydroxy-compound (from 7 g. of crotonate) was boiled with phosphorus oxychloride (30 ml.) for 15 min. and the solution was then evaporated under reduced pressure. The residue was extracted thrice with hot water, then filtered (charcoal), and the combined extracts were basified with 2N-sodium hydroxide. The amorphous chloro-compound (3.5 g.) which separated was purified by dissolving it in alcohol and slowly precipitating it with ether (Found : N, 12.4; Cl, 7.9. C₂₅H₂₃O₂N₄Cl requires N, 12.5; Cl, 7.95%). The picrate, m. p. 210° (decomp.), was sparingly soluble in alcohol (Found : N, 15.7. C₂₅H₂₃O₂N₄Cl,2C₆H₃O₇N₃ requires N, 15.5%). Evaporation of the alcohol-ether mother-liquors and hydrolysis of the residue with hydrochloric acid gave the hydrochloride of (IV; R = R' = H, R'' = Cl) (1.1 g.) (see below).

Reaction of 4-Acetamido-6-[N-(4-chloroquinald-6-ylmethyl)acetamido]quinaldine with Alcoholic Ammonia.—The 4-chloro-compound (1.2 g.) and saturated alcoholic ammonia (30 ml.) were heated for 7 hr. at 120° in a sealed tube. The resulting solution was evaporated and the residue crystallised from methanol, giving the chloro-amine (0.35 g.), pale yellow needles, m. p. 256° (Found : C, 65.2; H, 5.6; N, 13.25; Cl, 8.3; H₂O, 5.0. $C_{23}H_{21}ON_4Cl, H_2O$ requires C, 65.4; H, 5.5; N, 13.25; Cl, 8.4; H₂O, 4.3%). Some starting material was recovered from the mother-liquors.

4-Amino-6-[(4-chloroquinald-6-ylmethyl)amino]quinaldine.—The diacetyl compound (IV; R = R' = Ac, R'' = Cl) (1.5 g.) in 2N-hydrochloric acid (60 ml.) was heated on the steam-bath for 1 hr., then concentrated and cooled. The *dihydrochloride*, which separated, was recrystallised from N-hydrochloric acid; it had m. p. >360° (Found : C, 55.6; H, 5.2; N, 12.1; total Cl, 23.25; Cl⁻, 15.6. C₂₁H₁₉N₄Cl,2HCl,H₂O requires C, 55.6; H, 5.2; N, 12.1; total Cl, 23.5; Cl⁻, 15.6%).

4-Amino-6-[(4-aminoquinald-6-ylmethyl)amino]quinaldine.—A solution of the crude diacetyl compound (IV; R = R' = H, R'' = Cl) (4.45 g.) in phenol (25 g.) was heated at 185—195° for 3 hr., while a rapid stream of ammonia was passed through the solution. After being cooled, the solution was mixed with concentrated hydrochloric acid (20 ml.) and steam-distilled to remove phenol and to complete hydrolysis. The *dihydrochloride*, which separated on cooling, was recrystallised by dissolution in boiling water (100 ml.) cooling, and addition of concentrated hydrochloric acid. The yield was 2 g. (Found: C, 58·1; H, 6·7; N, 16·3; Cl, 16·5. $C_{21}H_{21}N_5$,2HCl,H₂O requires C, 58·0; H, 5·8; N, 16·1; Cl, 16·3%).

Condensation of 4:4'-Diaminostilbene with Acetoacetic Ester.—trans-4:4'-Diaminostilbene (13 g.) and acetoacetic ester (17 ml.) in methanol (100 ml.) containing concentrated hydrochloric acid (1 drop), were heated under reflux for 2 hr., then cooled and filtered to give the crotonate, $24\cdot2$ g. (90%), m. p. 177—184°, raised by recrystallisation from methanol to 184—185° (Found : C, 71·9; H, 7·0; N, 6·45. $C_{28}H_{30}O_4N_2$ requires C, 72·0; H, 6·9; N, 6·5%).

1: 2-Di-(4-hydroxyquinald-6-yl)ethylene.—The foregoing crotonate (39·1 g.) was slowly added to stirred boiling Dowtherm (200 ml.). After 15 min. the suspension was cooled and filtered, and the crude quinaldine (33·05 g.) was washed with benzene and used without further purification.

1: 2-Di-(4-methoxyquinald-6-yl)ethylene.—The crude hydroxy-compound (23 g.) and methyl sulphate (18.6 g.) were heated in dry toluene (80 ml.) for 3 hr. After being cooled, the product was filtered off, washed with benzene, and repeatedly extracted with boiling water (charcoal) until the extracts no longer gave a precipitate with alkali. The combined extracts were basified with concentrated aqueous sodium hydroxide and the precipitate was filtered off, washed with water, and crystallised from alcohol. The bright yellow methoxy-compound (8.8 g.) had m. p. 269--270° (Found : C, 77.5; H, 6.5; N, 7.5. $C_{24}H_{22}O_2N_2$ requires C, 77.8; H, 6.0; N, 7.6%). The yellow dihydrochloride, crystallised from N-hydrochloric acid and from aqueous alcohol, had m. p. >370° (Found : N, 5.9; Cl, 14.0. $C_{24}H_{22}O_2N_2$,2HCl,3H₂O requires N, 5.65; Cl, 14.25%). The bismetho(methyl sulphate), crystallised from methanol-ether, had m. p. 270° (decomp.) (Found : N, 4.55. $C_{24}H_{22}O_2N_2$,2Me₂SO₄ requires N, 4.5%).

1: 2-Di-(4-aminoquinald-6-yl)ethylene.—1: 2-Di-(4-methoxyquinald-6-yl)ethylene (3 g.) and anmonium acetate (30 g.) were heated together at 135—140° for 3 hr. The cooled mixture was dissolved in water, and excess of concentrated aqueous sodium hydroxide added. The diamine was filtered off, washed with water, and crystallised from alcohol (2·1 g.; m. p. >360°) (Found : C, 74·8; H, 6·3; N, 14·6. $C_{22}H_{20}N_4, C_2H_6O$ requires C, 74·6; H, 6·7; N, 14·5%). The dihydrochloride was purified by the addition of alcohol to its solution in the minimum quantity of hot water; it had m. p. >370° (Found : N, 11·5; Cl, 15·2. $C_{22}H_{20}N_4, 2HCl, 3·5H_2O$ requires N, 11·8; Cl, 14·9%). The bismetho(methyl sulphate), similarly purified from aqueous alcohol, was hygroscopic and had m. p. >295° (Found : N, 8·15. $C_{22}H_{20}N_4, 2Me_2SO_4, 5H_2O$ requires N, 8·2%).

l: 2-Di-(4-chloroquinald-6-yl)ethylene.—The crude l: 2-di-(4-hydroxyquinald-6-yl)ethylene (14.5 g.) was boiled with phosphorus oxychloride (100 ml.) for l hr., excess of reagent then being distilled off *in vacuo*. The residue was cooled and cautiously treated with dilute sodium hydroxide solution until alkaline; the chloro-compound which separated crystallised from chloroform (charcoal) in pale yellow prisms, m. p. 274—276° (decomp.) (Found: N, 7.2; Cl, 19.2. $C_{22}H_{16}N_2Cl_2$ requires N, 7.4; Cl, 18.7%). The dihydrochloride (Found: N, 6.1; Cl, 31.3. $C_{22}H_{16}N_2Cl_2$,2HCl requires N, 6.2; Cl, 31.4%) was almost insoluble in water and alcohol.

Condensation of 1: 2-Di-(4-chloroquinald-6-yl)ethylene with Amines.—The chloro-compound was heated with a large excess of the amine (approx. 10 mol.) in a sealed tube under the conditions given in the Table. In the cases of methylamine and ethylamine, 33% solutions of the base in alcohol were used. The products were isolated by warming the contents of the tube with dilute sodium hydroxide solution, filtering off the solid, washing it thoroughly with water, drying it, and crystallising it from the appropriate solvent. In organic solvents all the 4-substituted amino-compounds thus obtained formed yellow solutions which exhibited a brilliant blue or violet fluorescence. The quaternary salts were prepared by boiling a solution or suspension of the base in toluene with a slight excess of methyl sulphate for the time stated in the Table. The yellow or green products were purified from alcohol or methanol-ether.

1: 2-Di-(4-phenoxyquinald-6-yl)ethylene.—The 4-chloro-compound (4 g.) and phenol (10 g.) were heated under reflux for 3 hr. The resulting solution was basified with excess of sodium hydroxide solution, and the precipitate filtered off, washed with water, and crystallised from alcohol (charcoal) (yield 3.55 g., 68%; m. p. 250—255°). Recrystallisation gave 1: 2-di-(4-phenoxyquinald-6-yl)ethylene in brick-red prisms, m. p. 258—260° (Found : C, 81.4; H, 6.0; N, 5.8; H₂O, 1.4. C₃₄H₂₆O₂N₂,0.5H₂O requires C, 81.1; H, 5.4; N, 5.6; H₂O, 1.8%). The dihydrochloride (Found : N, 5.0; Cl, 12.9. C₃₄H₂₆O₂N₂,2HCl requires N, 4.8; Cl, 12.9%) formed a grey-green amorphous powder, soluble in cold water. The bismetho(methyl sulphate) (Found : N, 4.0. C₃₄H₂₆O₂N₂,2Me₂SO₄ requires N, 3.8%), prepared in the usual way, was a dark-blue solid, soluble in water to a mauve solution with a strong fluorescence.

Condensation of 2: 3-Di-p-aminophenylbut-2-ene with Acetoacetic Ester.—A solution of the amine (3.45 g.) and acetoacetic ester (4.3 g.) in methanol (20 ml.) containing concentrated hydrochloric acid (1 drop) was refluxed for 4.5 hr., then cooled overnight. The crotonate (5.9 g., 88%), m. p. 125°, was filtered off and recrystallised from methanol. It had m. p. 138° (softens at 134°) (Found : C, 72.85; H, 7.4; N, 6.35. $C_{28}H_{34}O_4N_3$ requires C, 72.7; H, 7.4; N, 6.05%).

2: 3-Di-(4-hydroxyquinald-6-yl)but-2-ene.—The foregoing crotonate (5.9 g.) was added to boiling Dowtherm (40 ml.). After 5 min. the suspension was cooled, diluted with benzene, and filtered and the crude quinaldine (4.65 g., 100%) washed with benzene and ether.

2:3-Di-(4-chloroquinald-6-yl)but-2-ene.—The crude hydroxy-compound (4.65 g.) and phosphorus oxychloride (30 ml.) were boiled for 30 min., excess of reagent then being distilled off *in vacuo*. The residue was cautiously treated with dilute aqueous sodium hydroxide, and the precipitate filtered off, washed, dried, and extracted with boiling benzene (charcoal). Evaporation of the extracts to a low bulk and addition of alcohol gave the *dichloro-compound* (1.75 g.), narrow plates, m. p. 232—234° (decomp.) (Found : N, 7.1; Cl, 17.0. $C_{24}H_{20}N_2Cl_2$ requires N, 6.9; Cl, 17.4%).

2: 3-Di-(4-aminoquinald-6-yl)but-2-ene.—The chloro-compound (1.75 g.) in phenol (15 g.) was heated at 190° and dry ammonia bubbled through the solution for 5 hr. The mixture was then steam-distilled to remove phenol and basified with aqueous sodium hydroxide, and the base filtered off, washed, and crystallised from aqueous ethanol. 2: 3-Di-(4-aminoquinald-6-yl)but-2-ene (1.35 g., 85%) formed prisms from chloroform or aqueous alcohol, decomp. approx. 308° (Found : C, 74.9; H, 6.4; N, 14.9. C₂₄H₃₄N₄, H₂O requires C, 74.6; H, 6.8; N, 14.5%).

The dihydrochloride separated from hot water in prisms (Found : N, 12.4; Cl, 15.5. $C_{24}H_{24}N_4$, 2HCl, H₂O requires N, 12.2; Cl, 15.5%). The bismetho(methyl sulphate) formed prisms, decomposing above 350°, from aqueous alcohol (Found : N, 9.2. $C_{24}H_{24}N_4$, 2Me₂SO₄ requires N, 9.0%).

2: 3-Di-(4-methoxyquinald-6-yl)but-2-ene.—Crude 2: 3-di-(4-hydroxyquinald-6-yl)but-2-ene (0.78 g.) and methyl sulphate (0.55 g.) in dry toluene (50 ml.) were boiled for 3 hr. then cooled. The solvent was decanted and the residual gum washed with ether, boiled with water (60 ml.) and charcoal, and filtered. 2: 3-Di-(4-methoxyquinald-6-yl)but-2-ene was precipitated by basifying the aqueous extracts, and formed needles (from aqueous alcohol), m. p. 240—242° (decomp.) (Found : C, 76.9; H, 6.3; N, 7.1; H₂O, 1.4. $C_{26}H_{26}O_2N_2, 0.33H_2O$ requires C, 77.1; H, 6.6; N, 6.9; H₂O, 1.5%).

The authors thank Dr. A. J. Ewins, F.R.S., for his interest, Dr. R. Wien and Mrs. R. Stone, B.Sc., for the trypanocidal tests, Mr. S. Bance, B.Sc., A.R.I.C., for the semimicroanalyses, and the Directors of May & Baker Ltd. for permission to publish these results.

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[Received, August 23rd, 1956.]